

A third polymorph of (2-thiazolidinethionato)-
(triphenylphosphine)gold(I)Ulrike E. I. Horvath* and
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Key indicators

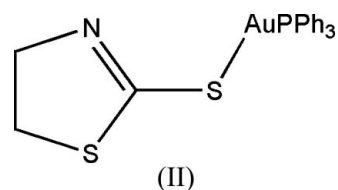
Single-crystal X-ray study
 $T = 100\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.011\text{ \AA}$
 R factor = 0.041
 wR factor = 0.088
Data-to-parameter ratio = 17.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $[\text{Au}(\text{C}_3\text{H}_4\text{NS}_2)(\text{C}_{18}\text{H}_{15}\text{P})]$, is a third polymorphic form of a compound previously reported by Grant, Forward & Fackler Jr [*Z. Kristallogr.* (1996). **211**, 483–484]. At 100 K, the present polymorph shows triclinic symmetry, as do the other two polymorphs.

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Comment

In the preceding paper (Horvath *et al.*, 2006), we reported a polymorph, (I), of a known compound (Grant *et al.*, 1996). The molecular structure of the title compound is shown in Fig. 1. The C1–S2 separation [1.693 (8) Å] in the thione ligand of (II) is somewhat less than in the other two polymorphic forms [1.733 (7) Å for (I) and 1.765 Å for the form reported by Grant *et al.* (1996)]. All the other bond lengths within the thione unit of the three structures are essentially identical. The influence of the Au atom and the non-rigid five-membered thione ring explains the rather larger thermal motion of atom S2, and also of C2 and C3. The most significant difference among the three polymorphs occurs in the distorted tetrahedral arrangement of the phenyl rings, with large differences in their respective torsion angles. This leads to a different packing mode of the present structure compared with those in the other polymorphs.



In contrast to polymorph (I), where the crystal structure is dominated by $\text{N}\cdots\text{H}$ and $\text{S}\cdots\text{H}$ contacts, leading towards a rather inefficient packing of the molecules, the packing along the a axis of (II) is now characterized by perpendicularly arranged molecules with $\text{N1}\cdots\text{H17}^i$ [symmetry code: (i) $x, y - 1, z$] contacts of about 2.58 Å with molecules of the neighbouring layer in the crystal structure (Fig. 2). The P1–Au1–S1 bond angle [176.06 (7)°] is nearly linear, as expected, and does not differ from reported values in similar structures (Macalindong *et al.*, 2006).

Experimental

The preparation of the compound followed a similar procedure as described for polymorph (I) (Horvath *et al.*, 2006). Single crystals suitable for X-ray diffraction were obtained by slow evaporation of a CH_2Cl_2 solution at 243 K.

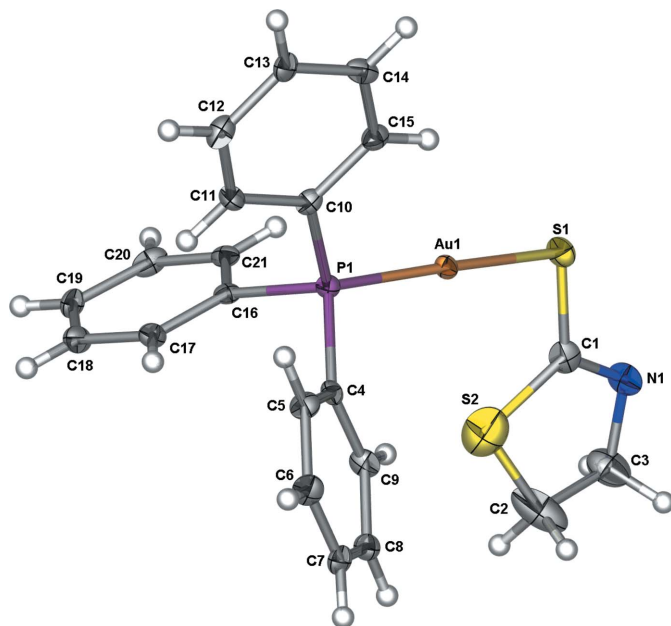


Figure 1
The molecular structure of (II), showing atom labels and 50% probability ellipsoids for non-H atoms.

Crystal data

$[\text{Au}(\text{C}_3\text{H}_4\text{NS}_2)(\text{C}_{18}\text{H}_{15}\text{P})]$	$V = 964.4 (2) \text{ \AA}^3$
$M_r = 577.43$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.988 \text{ Mg m}^{-3}$
$a = 8.2258 (11) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 11.0179 (15) \text{ \AA}$	$\mu = 7.93 \text{ mm}^{-1}$
$c = 12.1641 (17) \text{ \AA}$	$T = 100 (2) \text{ K}$
$\alpha = 71.411 (2)^\circ$	Block, colourless
$\beta = 75.984 (2)^\circ$	$0.14 \times 0.12 \times 0.12 \text{ mm}$
$\gamma = 68.848 (2)^\circ$	

Data collection

Bruker APEX CCD area-detector diffractometer	10322 measured reflections
ω scans	3924 independent reflections
Absorption correction: multi-scan (SADABS; Blessing, 1995)	3596 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.305$, $T_{\max} = 0.386$	$R_{\text{int}} = 0.041$
	$\theta_{\text{max}} = 26.4^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0333P)^2 + 4.2376P]$
$R[F^2 > 2\sigma(F^2)] = 0.041$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.088$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.11$	$\Delta\rho_{\text{max}} = 1.77 \text{ e \AA}^{-3}$
3924 reflections	$\Delta\rho_{\text{min}} = -2.15 \text{ e \AA}^{-3}$
223 parameters	
H-atom parameters constrained	

All H atoms were positioned geometrically (C–H = 0.95 and 0.99 Å for aromatic CH and CH₂, respectively) and constrained to ride on their parent atoms; $U_{\text{iso}}(\text{H})$ values were set at 1.2 times $U_{\text{eq}}(\text{C})$ for CH₂ and CH groups. Due to an unexpected high electron density around C19 the anisotropic displacement of the atoms in *para* posi-

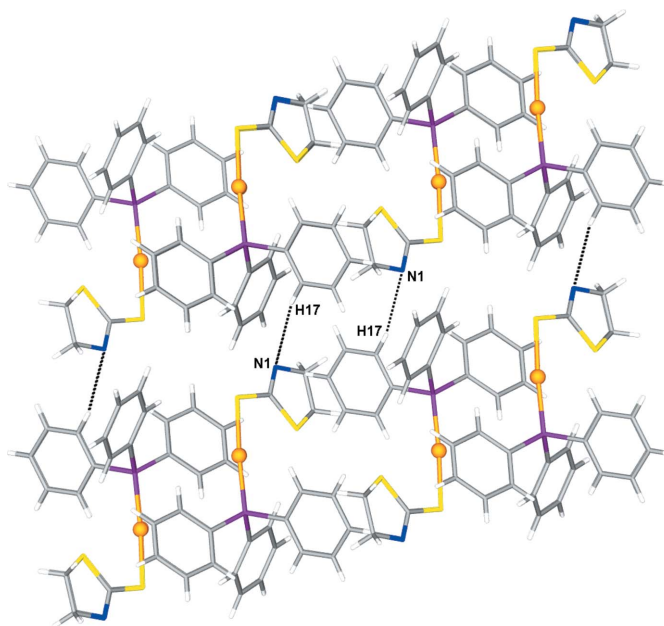


Figure 2
The packing, viewed along the a axis. The closest N...H contacts are shown as dashed lines. The symmetry code for atom H17 is $(x, y - 1, z)$.

tions were constrained to be the same as C7 using the EADP routine in *SHELXL97*. The highest peak is 0.52 Å from Au1 and the deepest hole is 0.51 Å also from Au1.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2002); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Atwood & Barbour, 2003; Barbour, 2001); software used to prepare material for publication: *X-SEED*.

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